IIT researchers develop materials to detect hydrogen gas leaks with high sensitivity



Dr. Chandra Shekhar Sharma, Associate Professor, Dept of Chemical Engineering, IIT-Hyderabad with Akash Nathani, Research Scholar.

Researchers at the Indian Institute of Technology Hyderabad and Indian Institute of Technology Jodhpur have developed materials that can detect hydrogen gas leaks with high sensitivity.

The research team has synthesized a semiconductor material that can be used as a sensitive detector of hydrogen gas. The work by the IIT-Hyderabad and IIT-Jodhpur team would help in the development of reliable and robust hydrogen gas sensors with high sensitivity and quick response, for domestic as well as industrial applications.

The results of the collaborative research have been published recently in the reputed peer-reviewed journal Sensors and Actuators.

The paper was co-authored by Chandra Shekhar Sharma, Associate Professor, Creative and Advanced Research Based On Nanomaterials (CARBON) Laboratory, Department of Chemical Engineering, IIT Hyderabad and Dr. Mahesh Kumar, Associate Professor, Department of Electrical Engineering, IIT Jodhpur, and research scholars, Vijendra Singh Bhati, Akash Nathani and Adarsh Nigam.

Dr. Sharma said that the importance of hydrogen has grown tremendously because of its promise as a primary energy source with the potential to become a panacea for clean energy generation. However,

there are two problems associated with the gas — it is difficult to produce and hard to detect, and because hydrogen is inflammable, leaks can be catastrophic if not detected immediately.

The researchers attempted to solve the latter problem through collaborative research.

Dr. Kumar explained that the explosive range (4–75%) and low ignition energy of hydrogen make leaks very dangerous. "Human beings can't sense hydrogen because it is colorless and odorless, which makes it essential to develop tools and sensors to detect hydrogen."

The authors said that many kinds of hydrogen sensors are being studied, including optical, electrochemical and electrical sensors. Electrical sensors, in particular, resistive sensors, are the closest to practicality due to their low cost, simple design and the possibility of good sensitivity.

Explaining how resistive sensors work, Dr. Kumar said: "Certain materials undergo a change in their electrical resistance when they come in contact with specific gases. For example, metal oxide semiconductors such as zinc oxide (ZnO) show changes in their electrical resistance when they come in contact with hydrogen."

"The sensitivity of ZnO's response to hydrogen gas depends upon its surface area - the larger the surface area, or the smaller the particles, the better its sensitivity," said Dr. Sharma.

Nanoparticles of ZnO, in which the particle sizes are a hundred thousand times smaller than the width of a single human hair, have good hydrogen sensing properties, but the research team has improved the sensitivity of this material even further.

The researchers have loaded the ZnO nanoparticles onto nanofibres of carbon and have shown that this results in a sensing response of nearly 74% compared to 44.5% in pure ZnO nanoparticles. Nanofibres are extremely thin fibres that are bundled to look like cotton candy.

The team attributes this improvement in performance to the easy diffusion of hydrogen gas through the nanopores of the cotton-candy-like carbon nanofibres, thus bringing them into intimate contact with the ZnO nanoparticles that are deposited on the nanofibres.

The team spun the nanofibres by a process called electrospinning, in which a polymer solution is electrically charged and ejected through a spinneret under a high-voltage electric field. In this work, the researchers used a special polymer blend to obtain nanofibres of the polymer, which was then converted to carbon nanofibres by heating.

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